



CREATING
A CIRCULAR
FUTURE

Assessing impacts of over fifty years of launching activities on marine ecosystems (the metal compounds case)

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Université de Montpellier

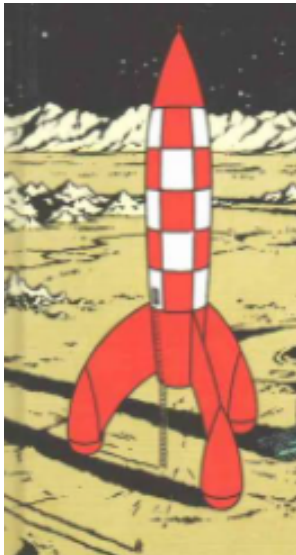


Deloitte.

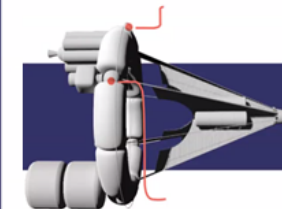
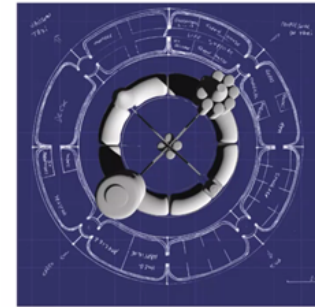
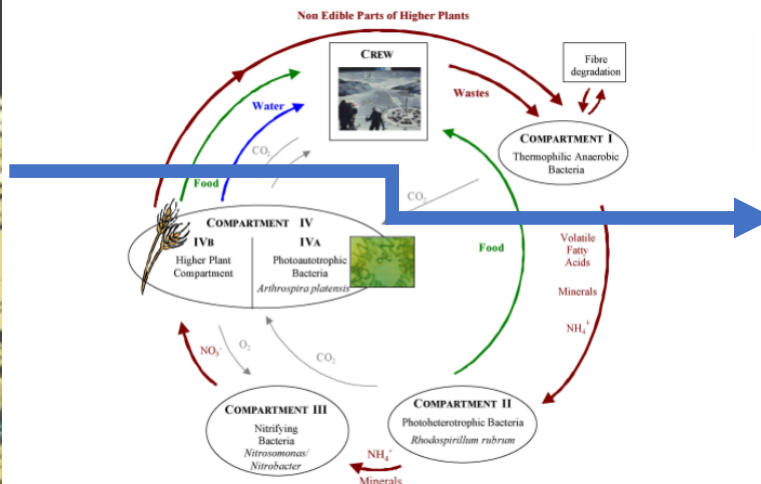




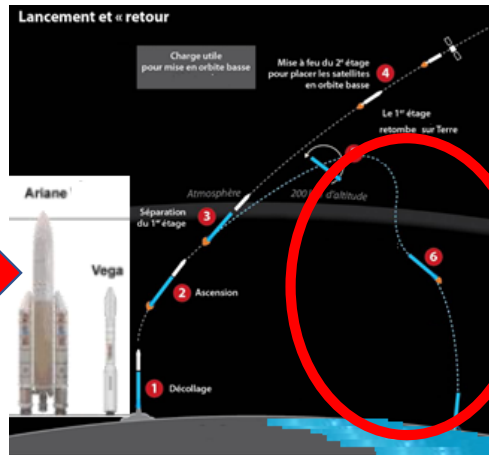
The way toward a « circular future »?



Source : Hergé Editions Castelman 1953



Source : didier schmitt, this conference
copyright: D. Schmitt, Red Safari, Editions Weyrich



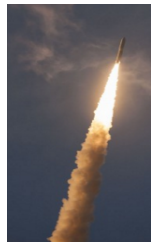
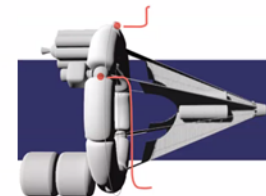
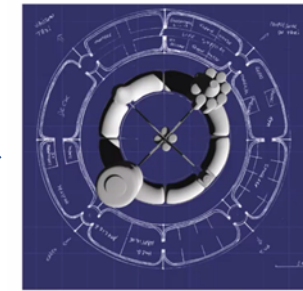
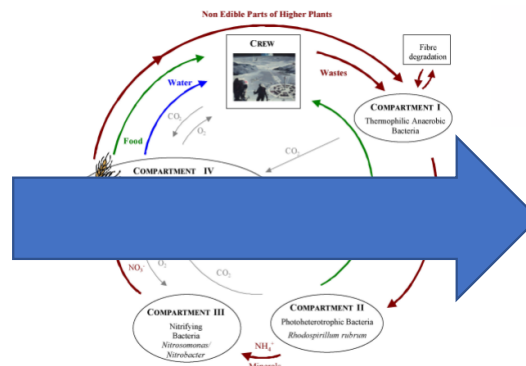
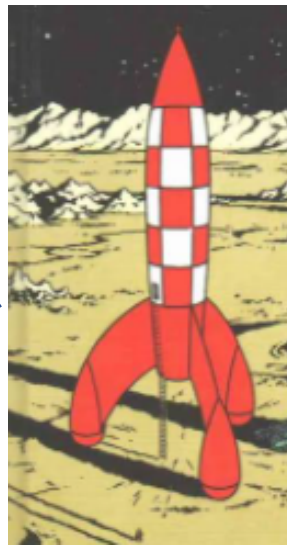
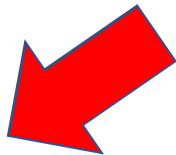
Since 2008 the Marine Strategy Framework Directive (2008/56/EC) of the European Parliament imposes to preserve and protect the marine environment

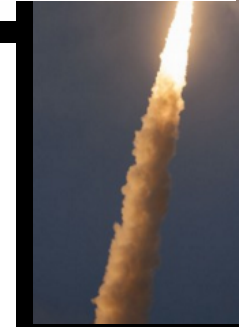
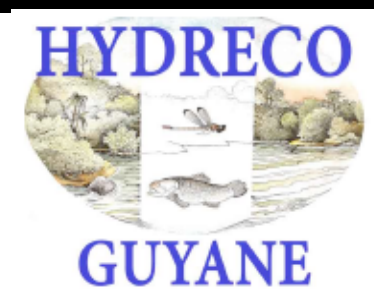
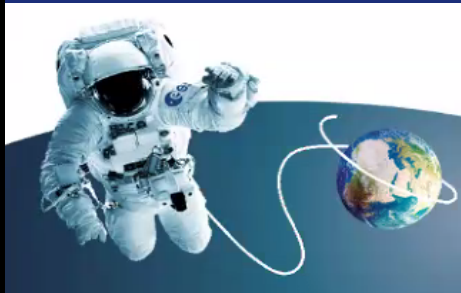
Since 2012, the space sector initiated the Clean Space initiative to investigate Life Cycle of Europe's launch vehicles and missions.

up to now ecotoxicological impacts on marine ecosystems of launching, and particularly, of launching residuals were under investigated and oversimplified.

Our objectives were:

- To help ESA to face the increasing concern of its customers, stakeholders and the general Public about environmental impact of launching
- To give ESA the possibility to progressively move toward ecodesign – by identifying key risk materials and suggesting design/process changes that could possibly reduce environmental impact of ESA launchers.





Functional Ecology to Reduce Launchers Impact on Deep Sea (ESA AO 1-8623/16/NL/KML)



ESTIMATED RELEASE to the SEA



INDUSTRY

300-500 billion kg/year
(heavy metals, sludges,
solvents..)

ESTIMATED RELEASE
to the SEA
12700 kg/sec



SUN SCREEN

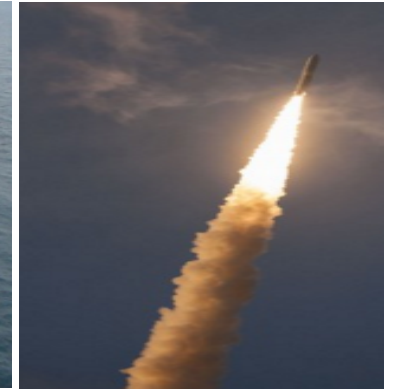
25000 t/year
ESTIMATED RELEASE
to the SEA
0,8 l/sec



SHIPPING

1million t /year
(gaz discharge)

ESTIMATED RELEASE
to the SEA
0,76 kg/sec



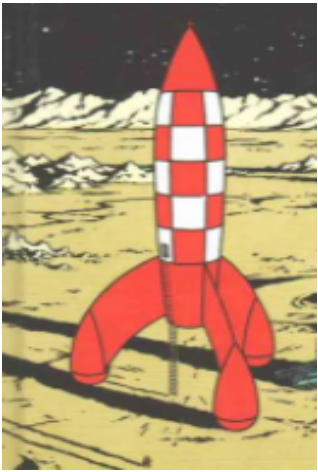
LAUNCHING

?

Source: Comparative analysis with main reported contaminations (from UNESCO data)

Global IMPACT assessments





Source : Hergé
Editions Castelman
1953



Source D. Schmitt, Red Safari
Editions Weyrich
2020

From
early stage

IMPACTS?

to 21st
century?



Inventory of launcher components



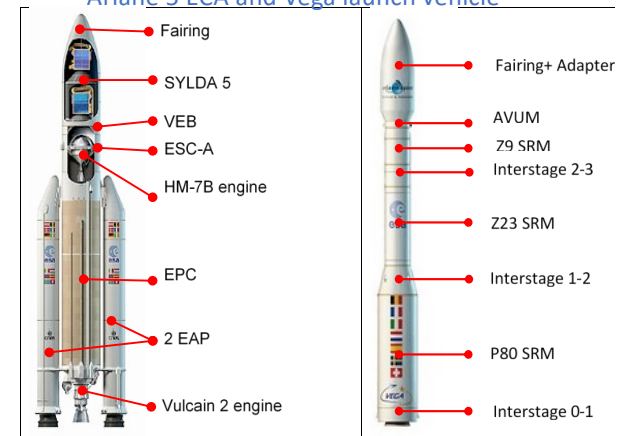
- Data sources:

- Ariane V and Vega LCA reports
- Documentation from ESA
- Literature review (internet search on chemistry websites, Material data sheets, Chemistry Handbook, etc.)
- Assumptions, either on the mass and/or size of certain components, or on the type/composition of material used

- The inventory was built in several steps:

- Determination of mass and surface for each component..
- Decomposition in chemical elements for complex materials.
- Compilation of physical-chemical characteristics of each chemical element and raw material from different data sources

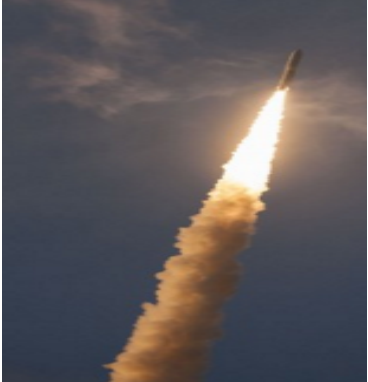
Ariane 5 ECA and Vega launch vehicle



Estimation of total mass of launching materials returning to earth

Materials and chemical elements	Mass per launchers (kg/launch)		
	A5 - ECA	A5 - ES	Sum of VEGA
Alumine	0,31	0,27	0,23
Aluminium	20852,18	19249,43	2605,18
Ammonium perchlorate	644,18	644,18	162,04
Antimony	96,70	96,70	0,00
Antimony trioxide Sb ₂ O ₃	0,08	0,06	0,05
Aramid fibre	125,40	125,40	500,50
B/KNO ₃ (pyrotechnic mixture)	2,40	0,00	0,00
Bentonite	73,35	73,35	0,00
BeO	0,00	0,00	0,00
Bismaleimide Triazine	0,00	0,01	0,01

ESTIMATED RELEASE to the SEA



INDUSTRY
 300-500 billion kg/year
 (heavy metals, sludges,
 solvents..)
ESTIMATED RELEASE
 to the SEA
 12700 kg/sec

SUN SCREEN
 25000 t/year
ESTIMATED RELEASE
 to the SEA
 0,8 l/sec

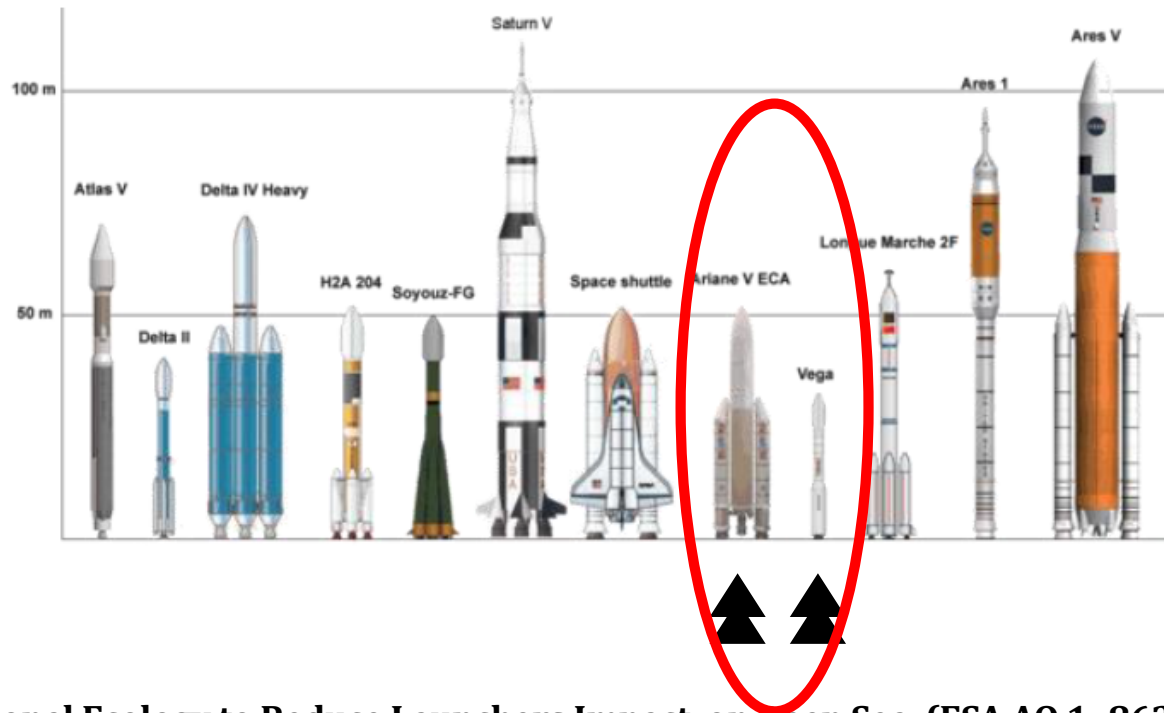
SHIPPING
 1million t /year
 (gaz discharge)
ESTIMATED RELEASE
 to the SEA
 0,76 kg/sec

LAUNCHING
ESTIMATED
RELEASE
 to the SEA
 14280 t/year
 0,45 kg/sec

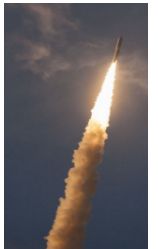
Source: Comparative analysis with main reported contaminations (from UNESCO data)

:

I-Launcher composition (ARIANE and VEGA) : (Eco) toxicity of components



Functional Ecology to Reduce Launchers Impact on Deep Sea (ESA AO 1- 8623/16/NL/KML)



<1
1-10
10-70
70-150
150-500
500-1000
1000-2000
2000-8000
>19 000

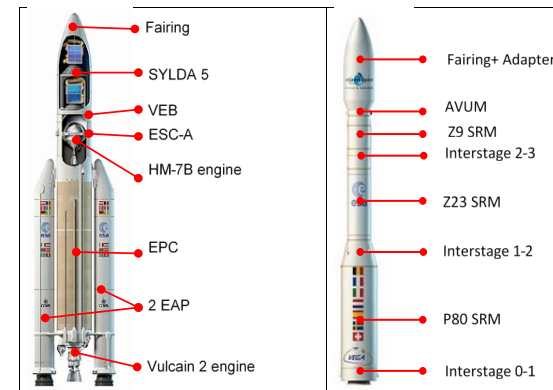
Substances with known toxicological or ecotoxicological hazards

List of LD50, LC50 and EC50 of substances



Component ^α	Mass ^α		Mammals ^α		Aquatic species ^α	
	A5 ^α	VEGA ^α	LD50-(mg/kg) ^α	LC50-(mg/L) ^α	EC50-(mg/L) ^α	
Alumina ^α	α	α	> 5000 (oral, rat) ^α	> 2.3 (4h, inhalation, rat) ^α	> 218.64 (96h, fish) ^α	
Aluminium ^α	α	α	2000 (mg/kg (oral, rat)) ^α	α	α	
Ammonium perchlorate ^α	α	α	1900 (oral, rat) ^α	α	α	
Antimony trioxide Sb2O3 ^α	α	α	34600 (oral, rat) ^α	α	> 1,000 (96h, fish) ^α	> 1,000 (daphnia magna) ^α > 67 (algae) ^α
Antimony ^α	α	α	7 (oral, rat) ^α	α	261 (24h, larvae) [¶] 4.92 (24h, flea) [¶] 206 (72h, algae) ^α	α
Aramid fibre ^α	α	α	7500 (oral, rat) ^α	α	α	
Bentonite ^α	α	α	35 (intravenous, rat) ^α	α	19000 (96h, fish) ^α	
BeO (beryllium oxide) ^α	α	α	2062 (oral, mouse) ^α	α	α	
Boron ^α	α	α	650 (oral, rat) ^α	α	α	
Bromine ^α	α	α	2.600 (oral, rat) ^α	2,700 (inhalation, rat) ^α	α	
CaCO3 (calcium carbonate) ^α	α	α	6450 (oral, rat) ^α	α	56 000 (48h, fish) ^α	
Cadmium ^α	α	α	2330 (oral, rat) ^α	α	0,001 (96h, fish) ^α	
Ceramic ^α	α	α	>22,500 (oral, rat) ^α	α	> 10 000 (72h, fish) ^α	
Chlorine ^α	α	α	α	293 ppm (1h, inhalation, rat) ^α	0 014 (96h, fish) ^α	
					0 019 (daphnia magna) ^α	

(Eco)Toxicity of launcher components



Our study showed that

- There were **30 substances** from the launchers with no **quantitative toxicological** information available.
- There **are 48 substances** from the launchers with no **quantitative ecotoxicological** information available.
- There are **26 substances** for which there are **neither toxicological nor ecological** quantitative information available
- For the two launchers, the top 5 of the **most risk substances were** : **paint, chlorine, cadmium, copper, zinc**

Alumina, mercury, ethylbenzene, nickel, sulphur are also part of the substances that can pose the highest risks to the marine environment



Metallic compounds



Literature review

Model experiments in laboratory conditions

On-site studies in French Guyana



Toxicity of components of launchers in marine conditions is function of their **composition** and **behavior**

However, previous data were obtained with the hypothesis of same substance release and same behavior (in terms of diffusion) in the ocean, while:

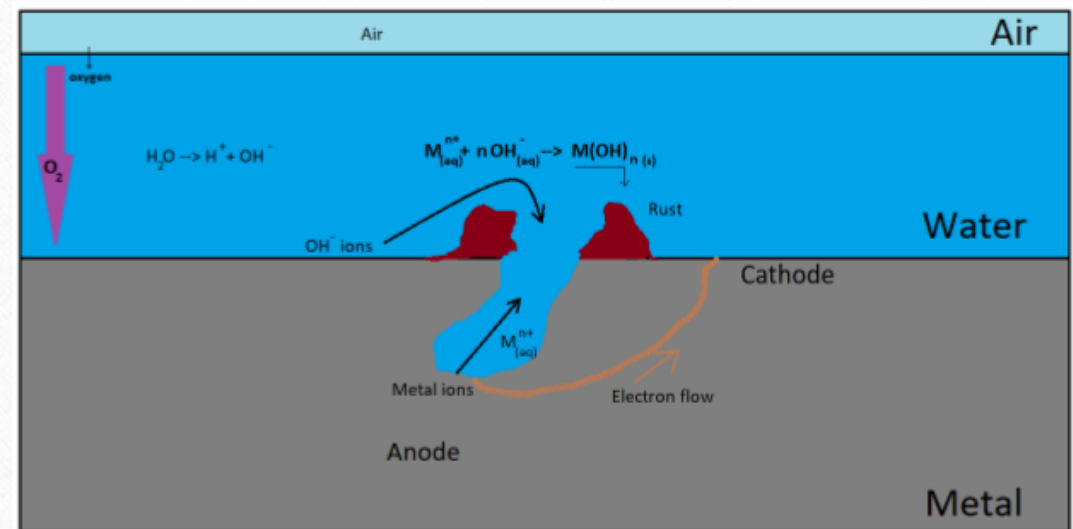
➔ Diffusion varies according to composition

➔ Diffusion varies according to environmental conditions



REACTIVITY = f(environmental conditions)

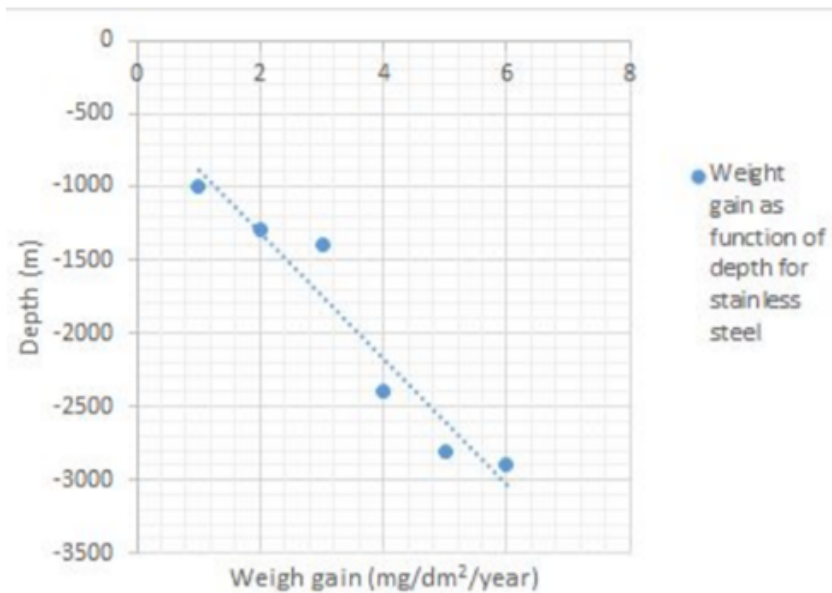
- oxydo-reduction
- Influencing parameters
 - salinity
 - depth
 - pH
 - dissolved O₂



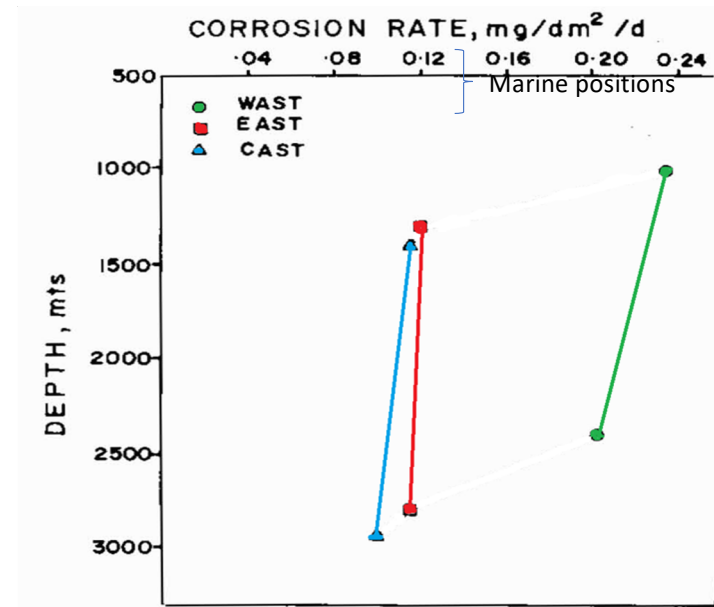
metal oxydation in sea water (KM3NeT Consortium, 2012)

Metal corrosion as a function of depth varies with the metal

Weight gain and corrosion rates as a function of depth for various metallic alloys (adapted from Sawant and Wagh, 1990)



! Inverse relationship for the corrosion rate of aluminium with depth



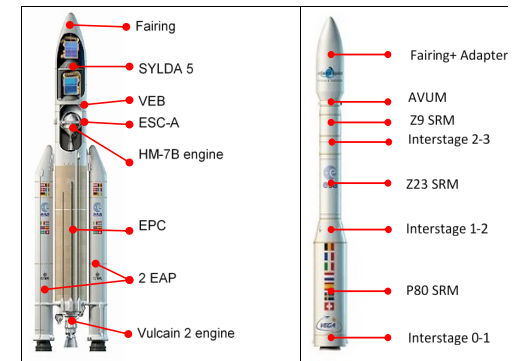
(adapted from Sawant and Wagh, 1990)

Metal (bio) corrosion in sea water = f (compounds composition)

metal	corrosion product in sea water	References
Fe	FeS ; Fe ₂ O ₃ ; Fe ₃ O ₄ ; Fe(OH) ₂ ; Fe(OH) ₃	Lee et Al. (2013) Byrne & Kester (1976)
Al	Al ³⁺ _(aq) ; Al(OH) ₄ ⁻ _(aq) ; Al(OH) _{3(aq)} ; Al(OH) ₂ ⁺ _(aq) ; AlO ₂ ⁻ _(aq) ; Al ₂ (SO ₄) _{3(s)} ; Al ₂ O _{3(s)} ; Al(OH) ₂ Cl ₂ ⁻	Sherif (2011) Wharton et Al. (2005)
Ti	TiO ; TiO ₂ ; TiO ₃ ; TiO _x N _y (TiN)	Totolin et Al. (2016)
Cu	Cu ₂ O ; CuO	Wharton et Al. (2005)
Cr	Cr ³⁺ ; Cr(OH) ₂ ⁺ ; Cr(OH) ₃ ; Cr(OH) ₄ ⁻ ; HCrO ₄ ⁻ ; CrO ₄ ²⁻ ; Cr ₂ O ₃	Van der Weijden & Reith (1982) Fukai (1967) Nakayama et Al. (1981)
V	VO ²⁺ ; H ₃ VO ₄ ; H ₂ VO ₄ ⁻ ; HVO ₄ ²⁻ ; VO ₄ ³⁻ ; VO ²⁺	Nukatsuka et Al. (2002)
Ni	Ni ²⁺ ; NiO	Van den Berg et Al. (1987) Gong et Al. (2011)
Zr	ZrO ²⁺	McKelvey & Orians (1993)
Mg	Mg(OH) ₂ ; Mg ²⁺ ; MgO	Zou et Al (2011) Pardo et Al. (2008)
Co	Co ²⁺	Ellwood & Van den Berg (2001)



Experimental set-up



- **metallic corrosion** in **seawater** and toxicity were followed in a range of conditions for a set of **representative alloys used in aerospace vessels**
 - Experimental set ups were designed for a set of **representative metal** components of VEGA and & ARIANE LAUNCHERS
 - Follow-ups were performed in **representative ecosystems** (French Guyana)

EXPERIMENTAL SET UP

- stainless steel and carbon steel
- sterile and non-sterile sea water
- 3 conditions
 - aerobiosis
 - anaerobiosis
 - interface



Selected steels

X15TN (stainless steel)

Witness

35NCD16 (carbon steel)

component of launchers



15CDV6 (carbon steel)

component of launchers



CHEMICAL COMPOSITION

	C	Si	Mn	Cr	Mo	V	N	Ni
min.	0.37	-	-	15.00	1.50	0.20	0.16	-
max.	0.45	0.60	0.60	16.50	1.90	0.40	0.25	0.30

COMPOSITION

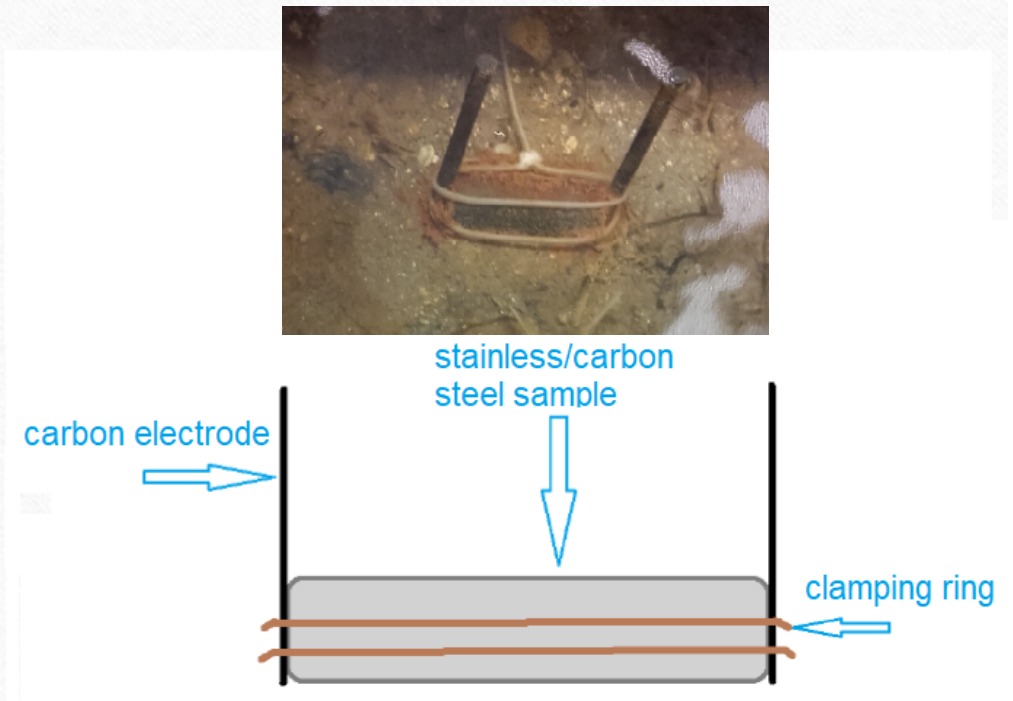
Carbone0,15
Chrome1,25
Molybdène0,90
Vanadium0,25

COMPOSITION

Carbone0,35
Nickel3,80
Chrome1,70
Molybdène0,30

Experimental set- up

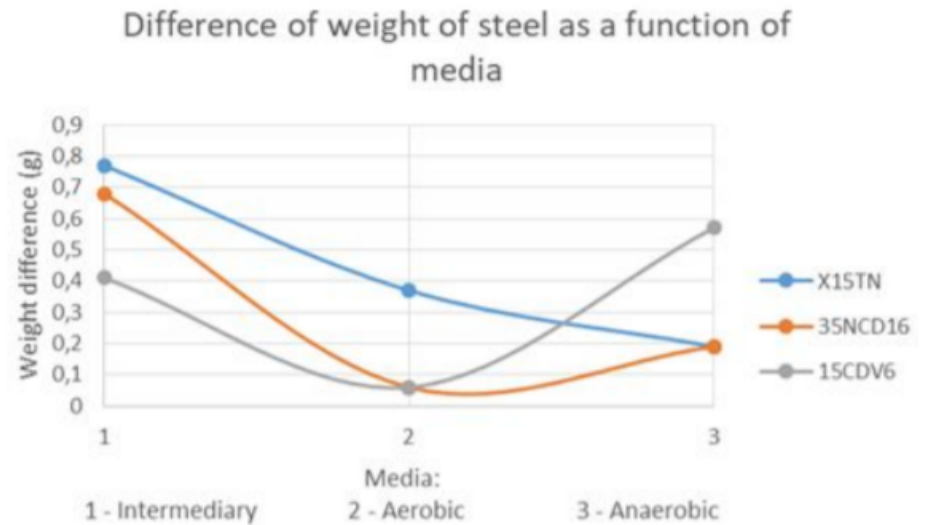
- pH follow-up
- measures of electric potential
- measures of sample weight
- electron microscopy observations
- ICP-MS elemental analysis
- ecotoxicological follow up



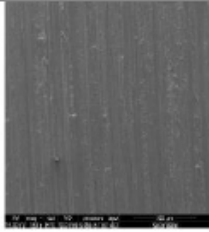
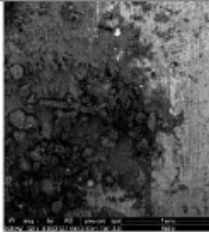
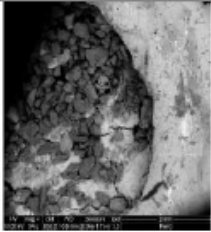
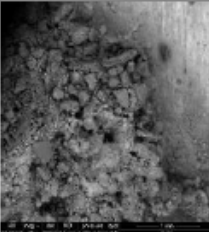
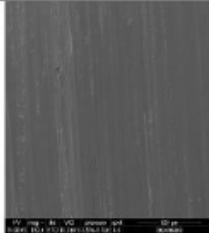
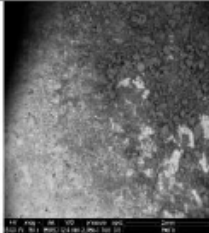
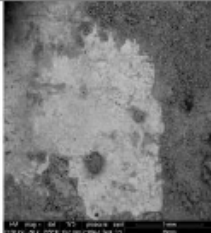
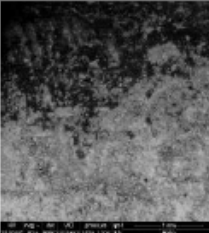
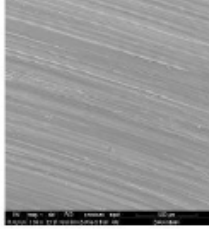
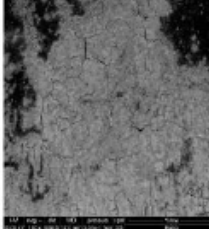

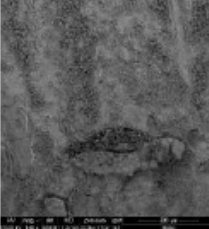
Metal corrosion as a function of composition and marine oxygen level

- Corrosion in seawater estimated from weight in aerobic, anaerobic and mixed conditions (our work, unpublished results)

Steel		Weight before	Weight after	difference
X15TN	Tank 1	120,36	119,59	0,77
	Tank 2	120,05	119,68	0,37
	Tank 3	120,08	119,89	0,19
35NCD16	Tank 1	122,88	122,2	0,68
	Tank 2	123,66	123,6	0,06
	Tank 3	122,89	122,7	0,19
15CDV6	Tank 1	122,7	122,29	0,41
	Tank 2	122,44	122,38	0,06
	Tank 3	122,56	121,99	0,57

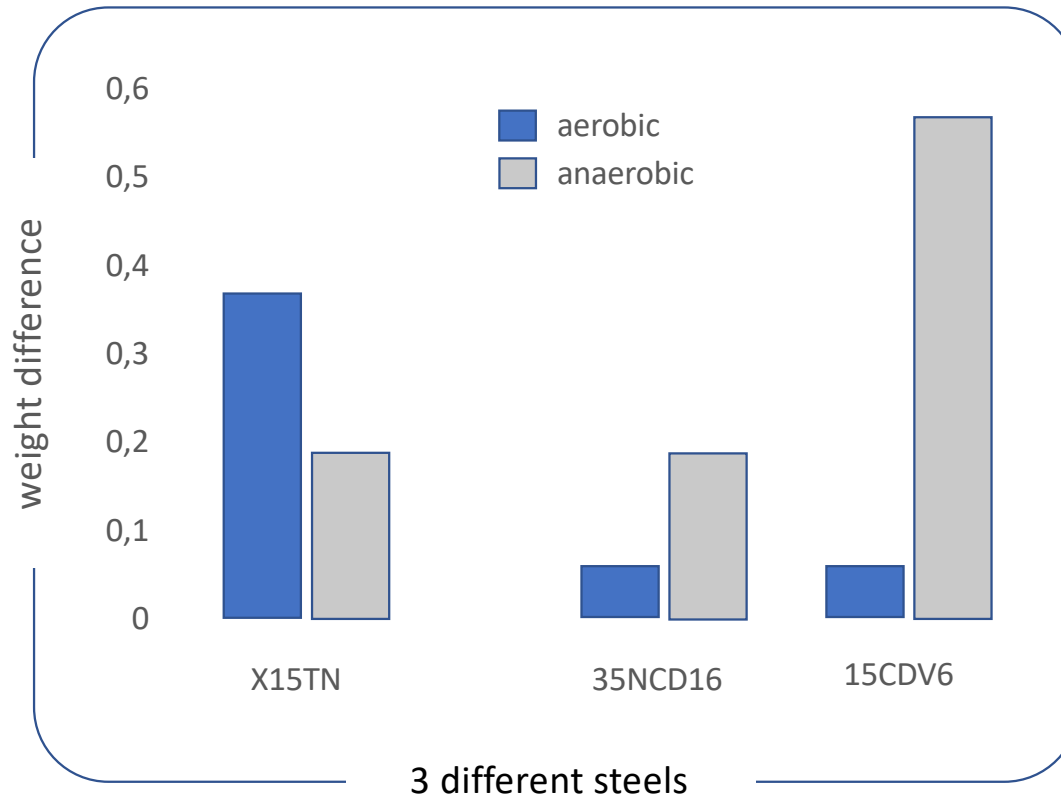


MEB analysis confirmed that (bio) corrosion depends on conditions

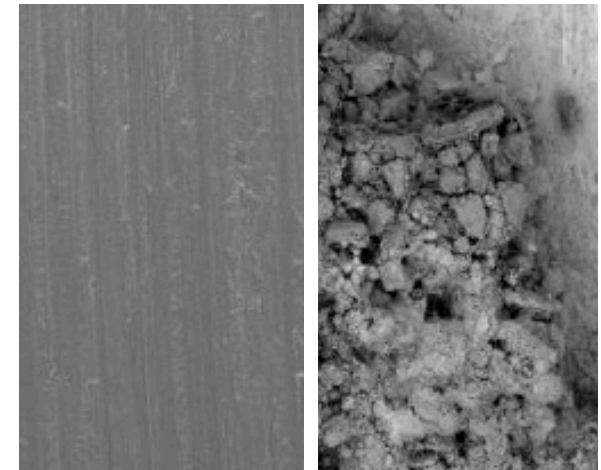
échantillons	Avant	Bac 1 après	Bac 2 après	Bac 3 après
X15TN				
35NCD16				
15CDV6				

Metallic corrosion as a function of metallic composition and seawater oxygen level

Corrosion in seawater estimated from weight in **aerobic** & **anaerobic** conditions

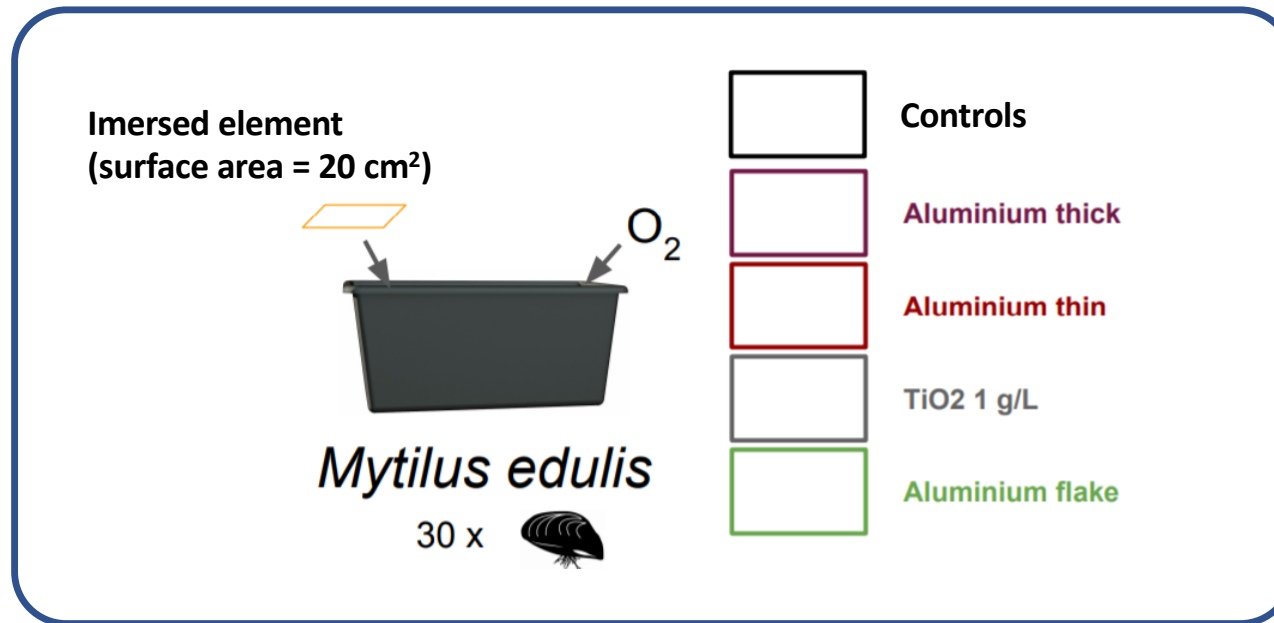


Surface pitting and cracks



Ecotoxicity evaluation

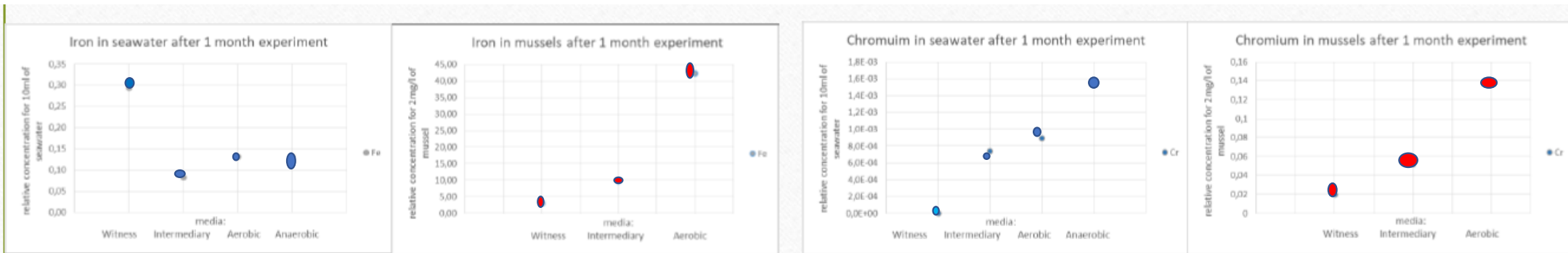
Macrofauna *in vitro*



Metal analysis (after 1 month) in sea water and mussels

Iron (in ppb)

Sea water: **stable** mussels : **increase**



Chromium (in ppb)

Sea water: **increase** mussels : **increase**

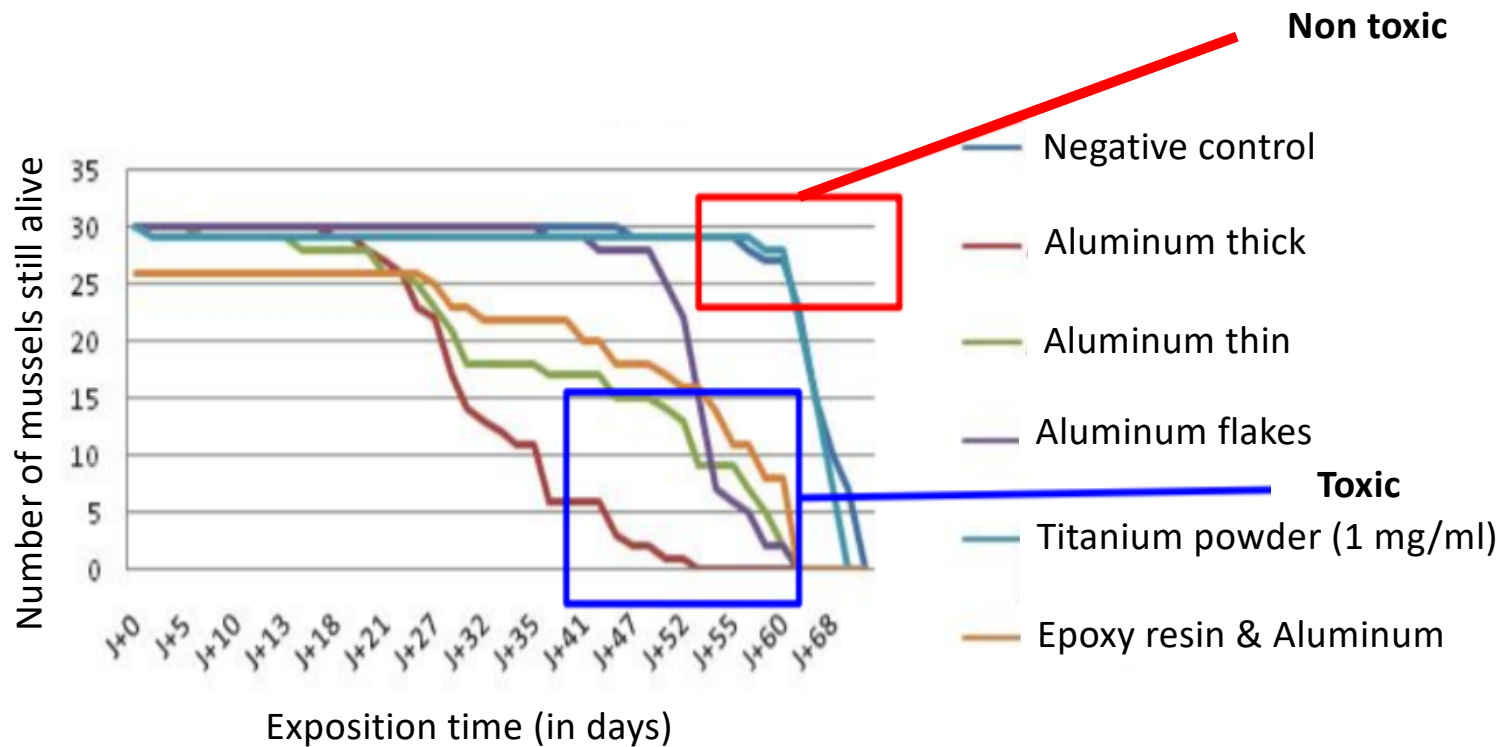
Graphiques représentant les concentrations en (ppb) de Fer dans l'eau de mer et dans les moules après un mois d'expérimentation.

Graphiques représentant les concentrations en (ppb) de Fer dans l'eau de mer et dans les moules après un mois d'expérimentation.

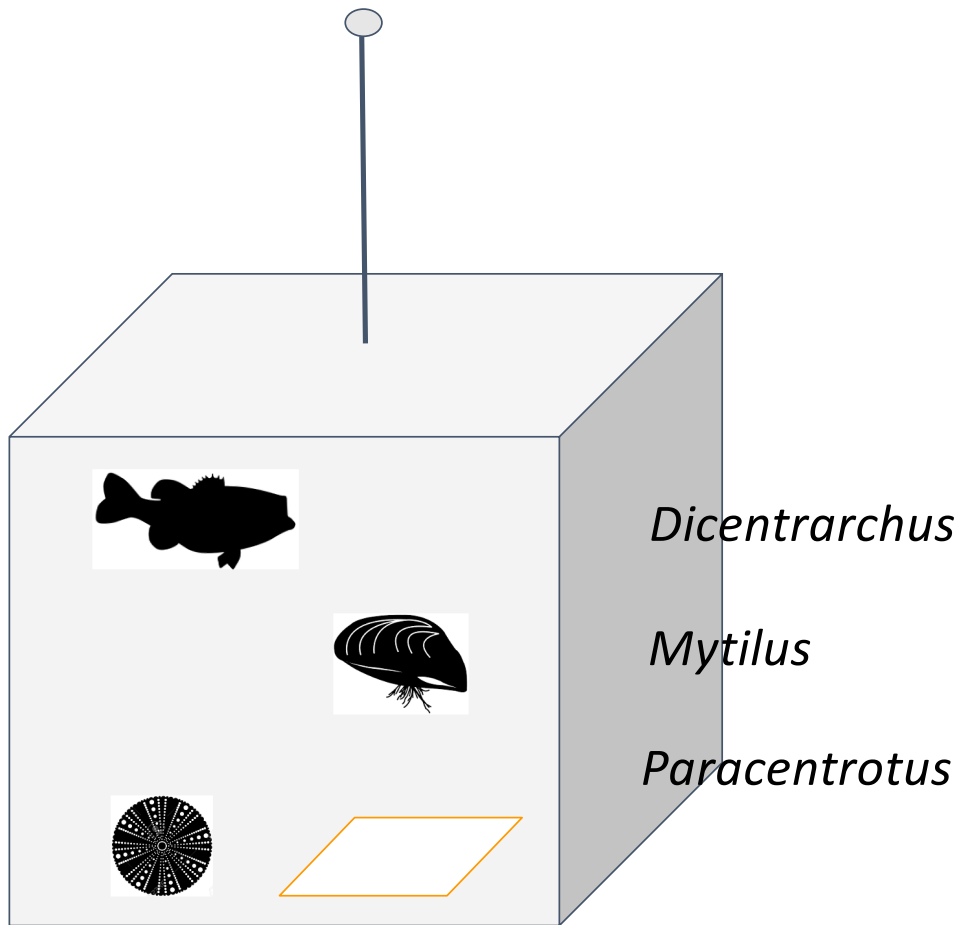
Ecotoxicity evaluation

Macrofauna *in vitro*

Survival curve for mussels exposed to different metals



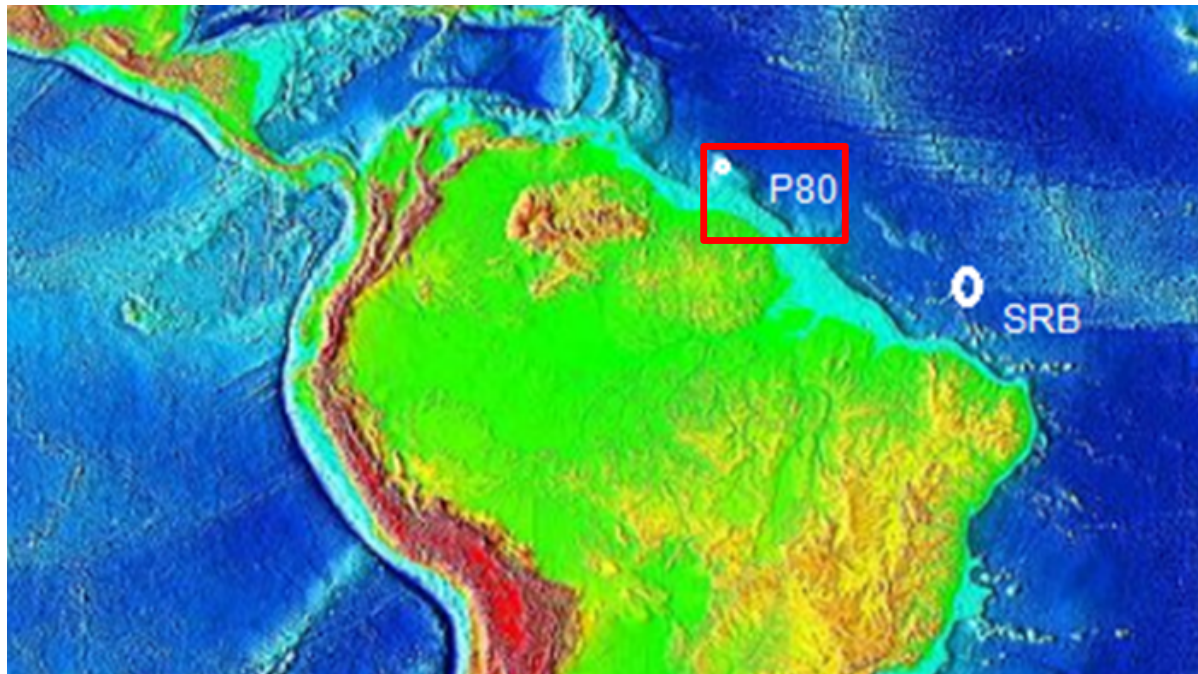
Ecotoxicological *'in situ'* assessments



IN SITU IMPACT STUDIES in French Guyana



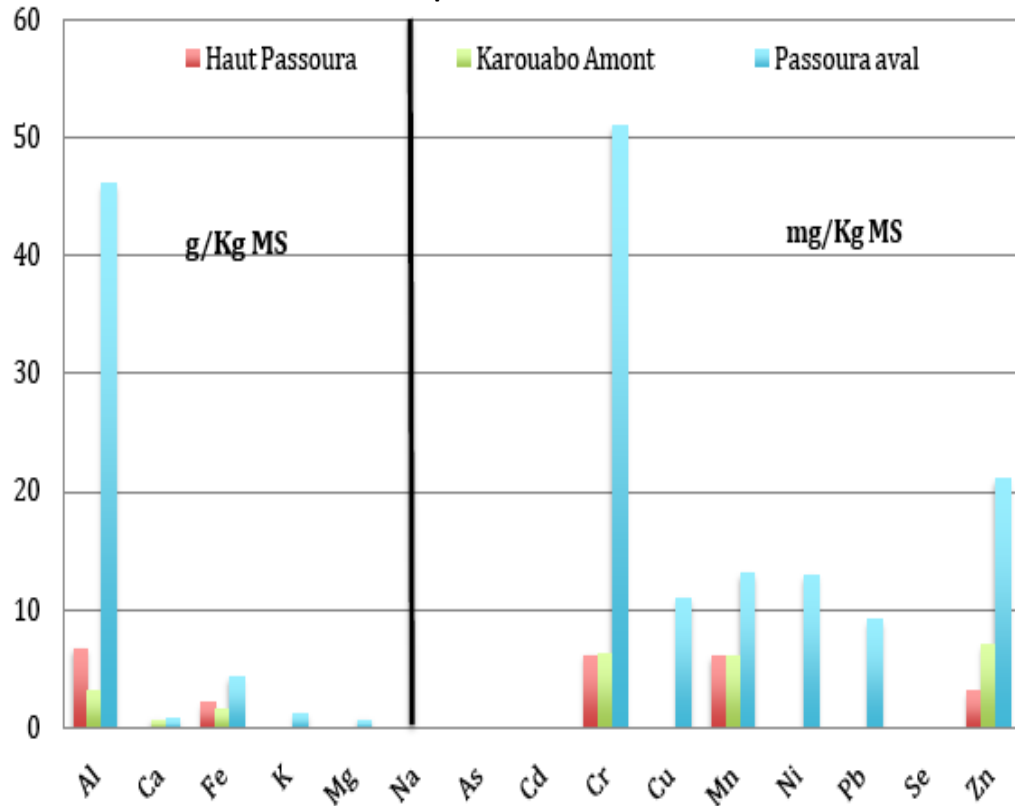
Ecotoxicological assessments near French Guyana first stage falling point



IN SITU IMPACT STUDIES in French Guyana

Sediment status

September 2014



In some locations, aluminum concentrations are high

Anthropogenic contaminations ?

IN SITU IMPACT STUDIES in French Guyana

Water Quality

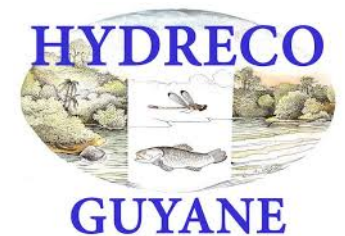
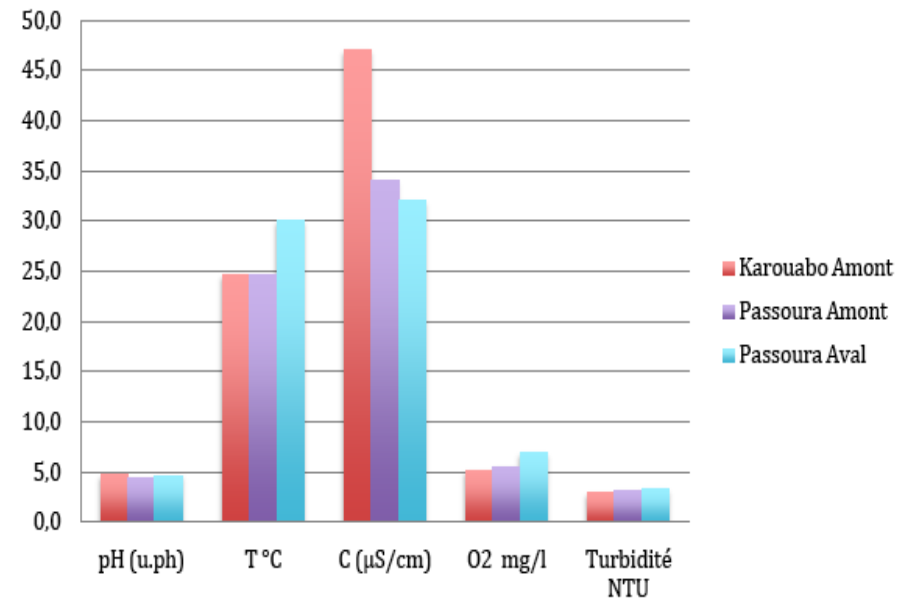
- Turbidity, Ph & Conductivity values are low
- Clear waters
- Physico-chemical characteristics of the environment change according to the season

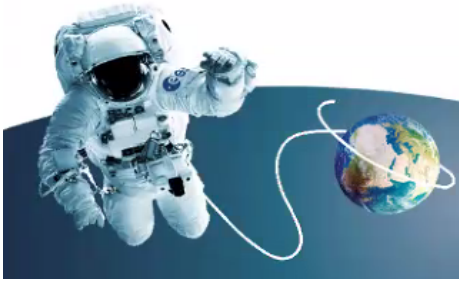
Fish populations

- Sites in good environmental health
- Strong imbalance in local fish population distribution (Vigouroux & Guillemet, 2006)

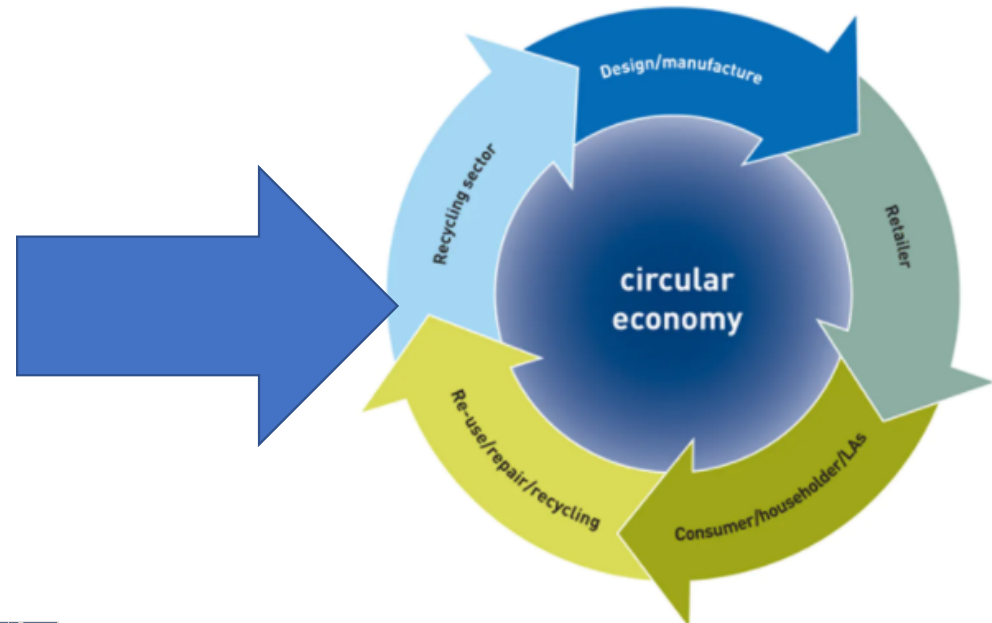
However, the successive launches of Ariane can induce the release of various products into the environment (including aluminum), with potentially a non-negligible impact on fish

In situ measurements





Conclusions & Perspectives



Conclusions & Perspectives

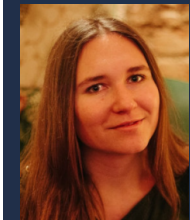
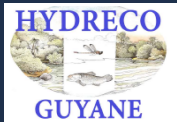
- Behaviour and toxicity of the different compounds carried out vary according to the substrate but also to environmental conditions (depth oxygen availabilities)
- Exposure time must be taken into consideration and no 'rapid' conclusion can be drawn from the available literature
=> experimental approach used during the project
- Longer impacts need to be investigated



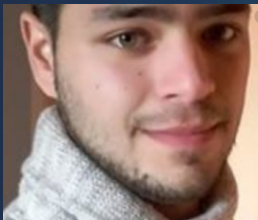
Regis
Vigouroux

Jan
Vanaverbeke

Steven
Degraer



Sophia
Klink



Eymeric
Fleouters



Benjamin
Lemoing



Jehan
Lignot



Deloitte.



Marianne
Planchon



Augustin
Chanoine

THANK YOU.

C. Lasseur



Stephanie
Raffestin

www.melissafoundation.org

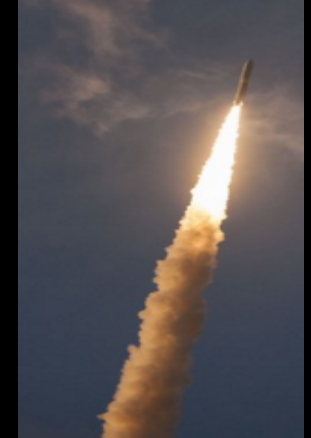
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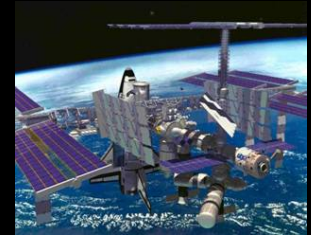


Functional Ecology to Reduce Launchers Impact on Deep Sea (ESA AO 1-8623/16/NL/KML)

Thank you for your attention !



UNESCO and France have formalized the creation in Montpellier of the International Centre for Interdisciplinary Research on Water Systems Dynamics (ICIReward) which will provide expertise, carry out research and training actions in management and governance, of water science and technology in vulnerable regions, focusing in particular on problems linked to rapid urbanization, demographic pressure and the foreseeable effects of climate change.



21.10.2020

