



Control of pH and process modelling contribute to stable alkalinity-limited urine nitrification

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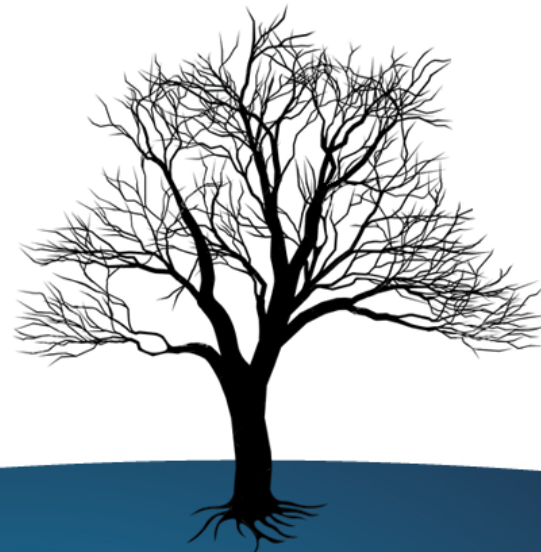
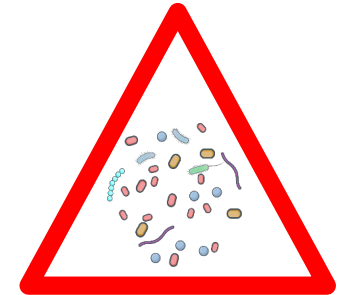
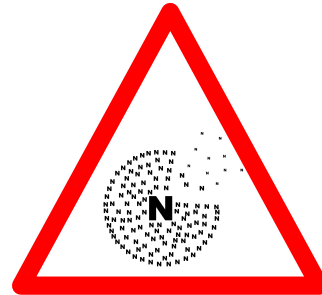


MELISSA: Urine treatment



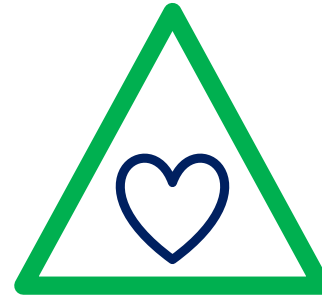
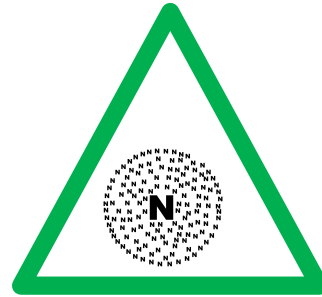


Why urine treatment?

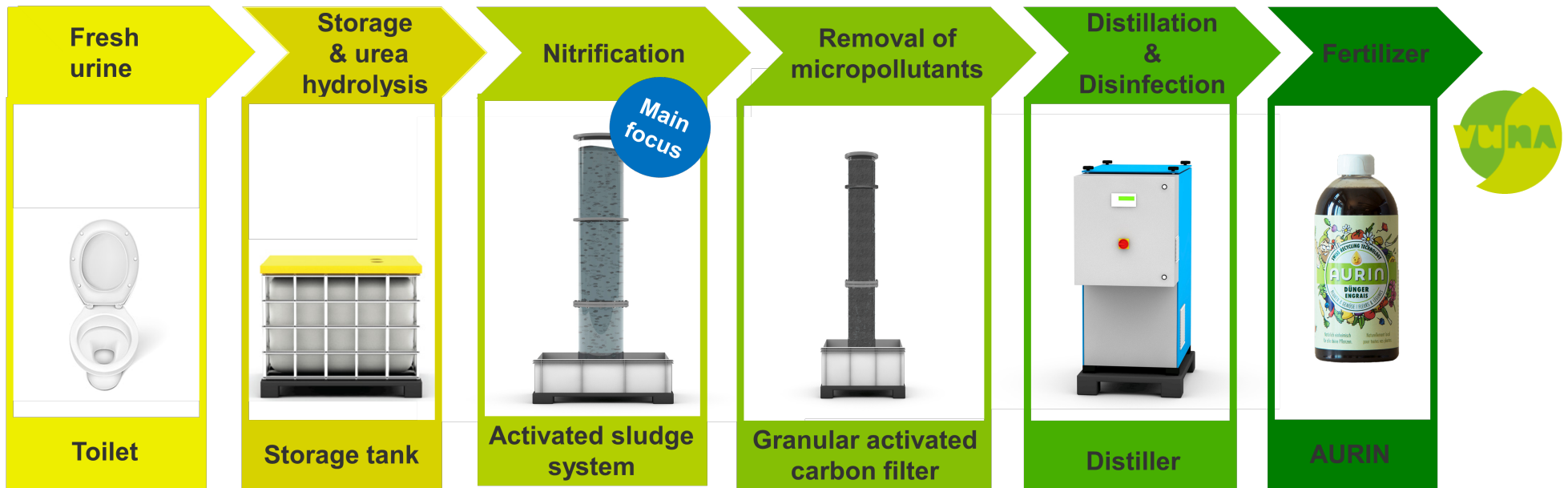




Why urine treatment?



Possible treatment chain



✗ Heavy metals (not present in urine)

Smell, volatile ammonia ✗

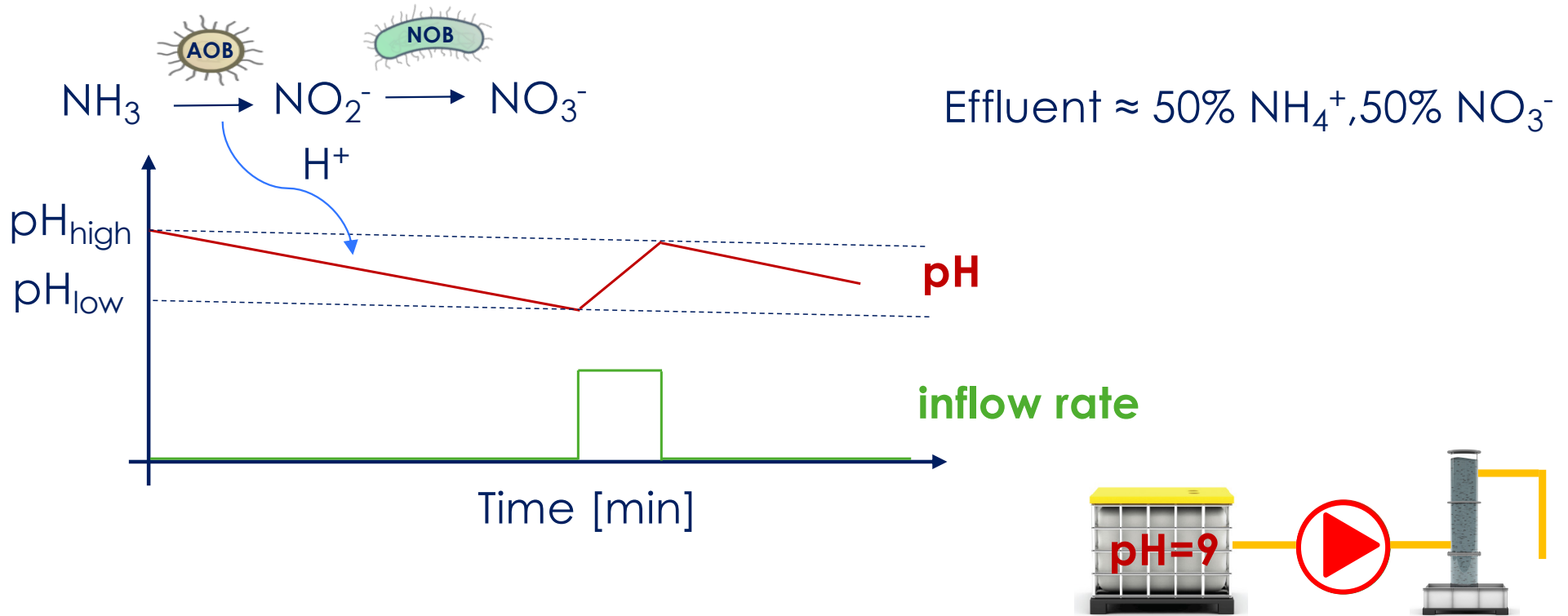
Pharmaceutical residues ✗

Pathogens ✗

Macro-nutrients (e.g. nitrogen, phosphorus, potassium) and micro-nutrients (e.g. zinc, boron) ✓



Alkalinity-limited urine nitrification





With or without alkalinity addition



Alkalinity-limited nitrification (50% NH_4^+ , 50% NO_3^-) Ex. Fumasoli et al. 2016	Full nitrification (100% NO_3^-) Ex. Jolien et al. 2020
<ul style="list-style-type: none">⊕ 50% less oxygen⊕ No base addition⊕ High TRL (licensed fertilizer)⊕ High urine treatment rate	<ul style="list-style-type: none">⊕ Rather stable process
<ul style="list-style-type: none">⊖ Process stability	<ul style="list-style-type: none">⊖ Oxygen consumption⊖ Base addition or electrolysis unit
<ul style="list-style-type: none">⊗ Ammonium-nitrate fertilizer	<ul style="list-style-type: none">⊗ Nitrate fertilizer⊗ TRL

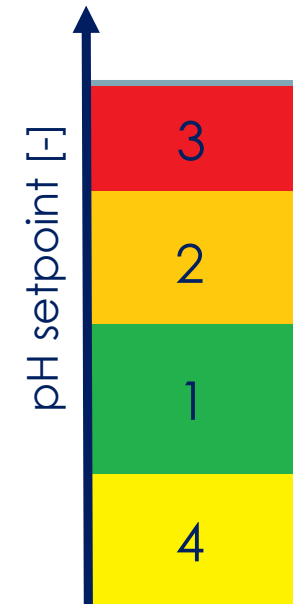


Process stability



Four different scenarios were observed in the past years:

1. Stable nitrification
2. Nitrite accumulation and nitritation
3. Complete stop of nitrification
4. Growth of acid-tolerant AOB





Hypothese



Goal of this study: Investigate the four different scenarios

1. pH=6 and pH=5.8 will lead to stable nitrification
2. pH=7 will lead to nitrite accumulation and nitritation
3. pH=8.5 will lead to a complete stop of nitrification
4. Inflow stop will lead to the growth of acid-tolerant AOB

Experimental procedure

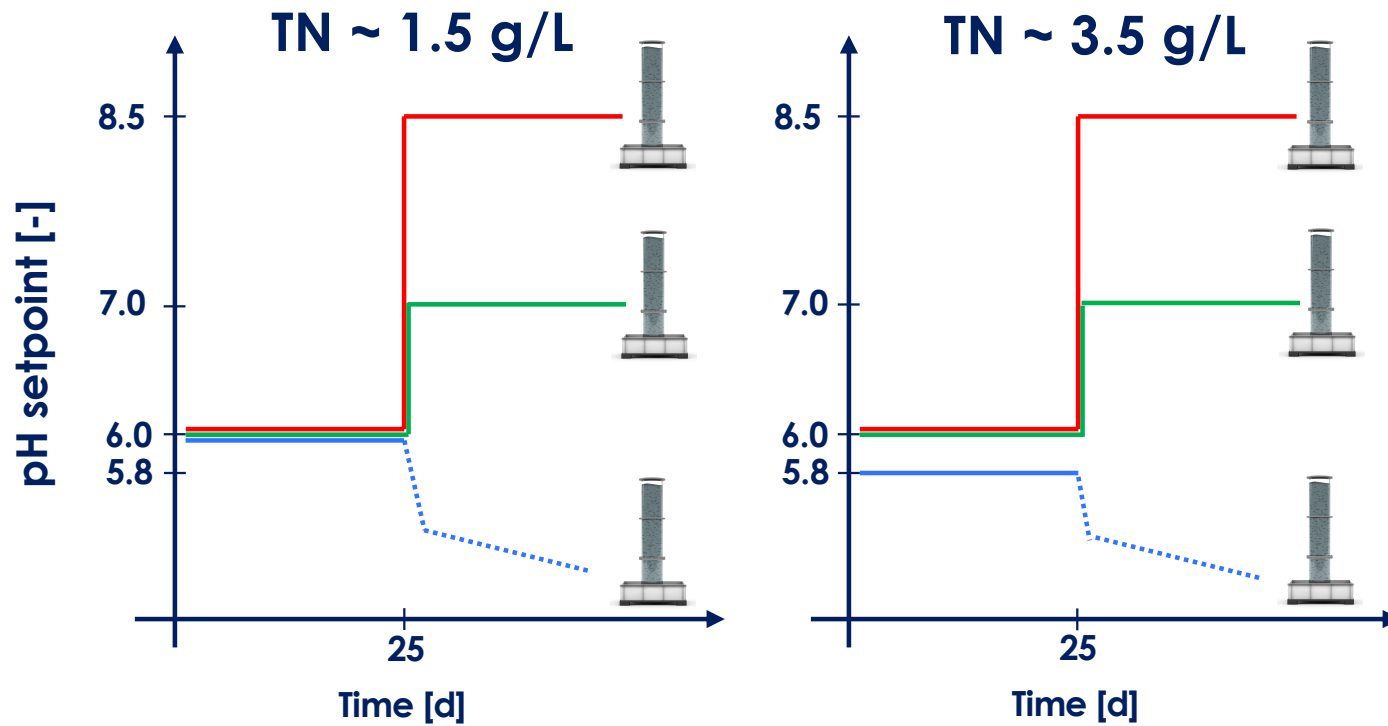
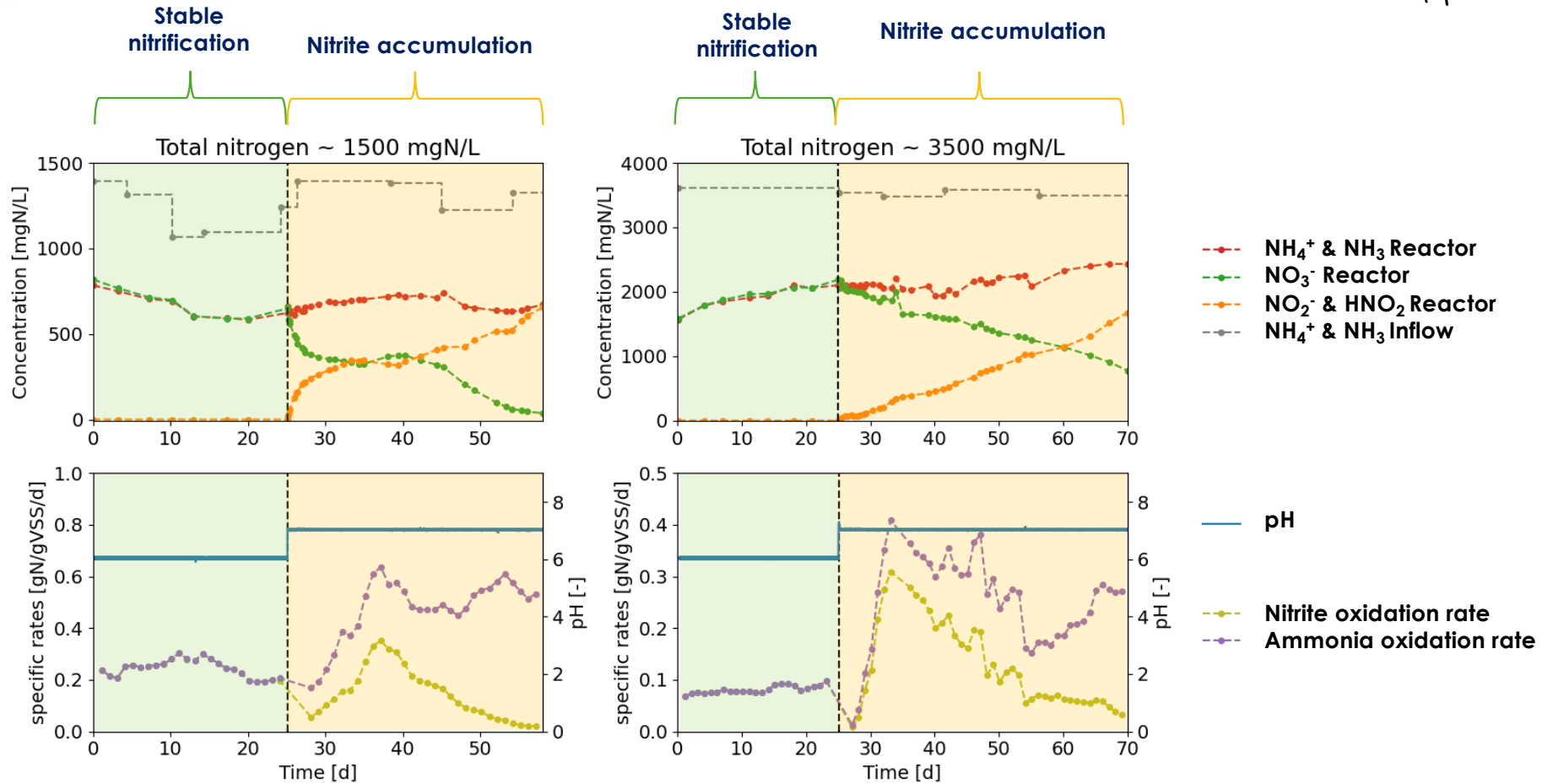
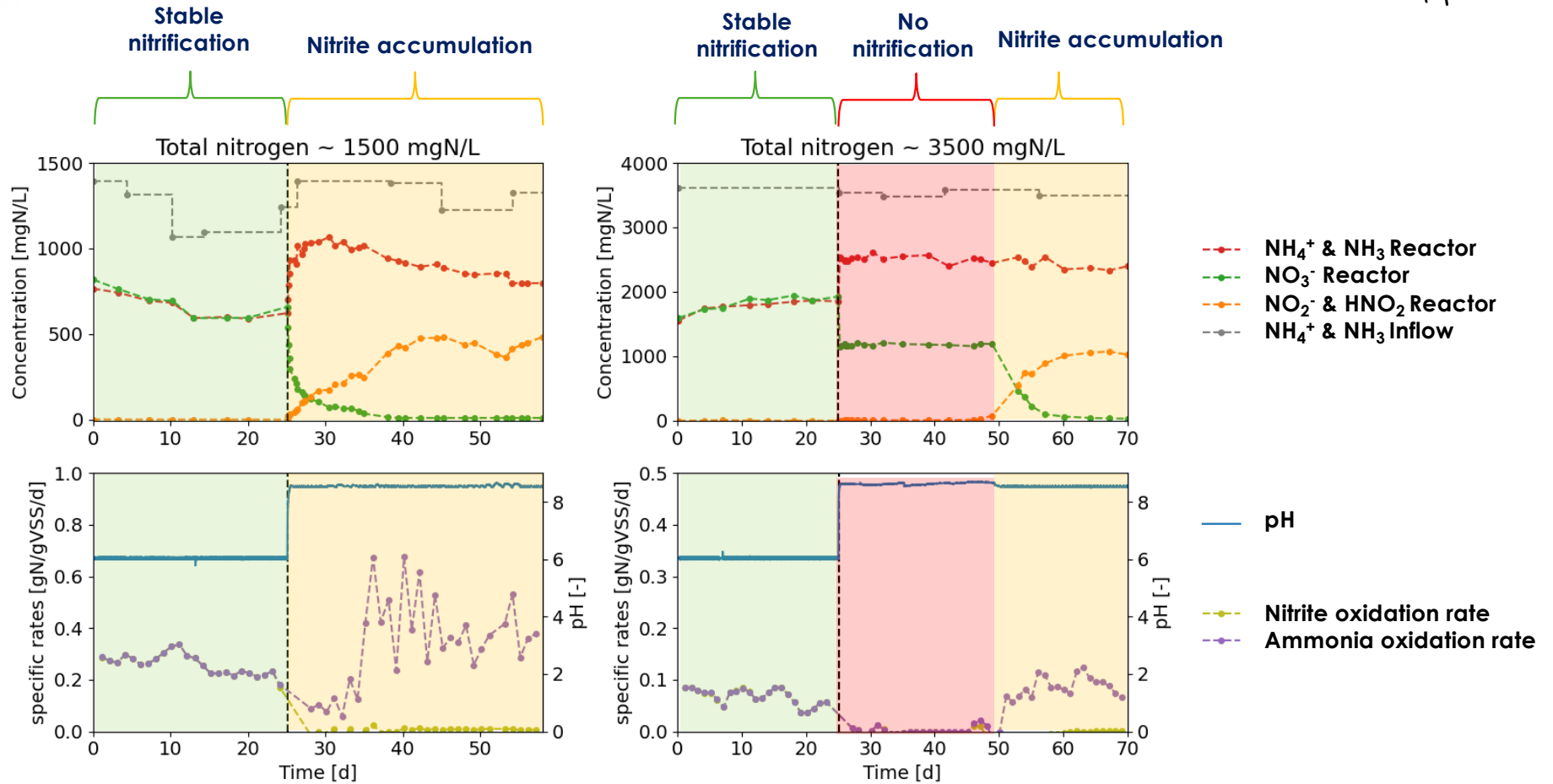


Fig.: 12L reactor

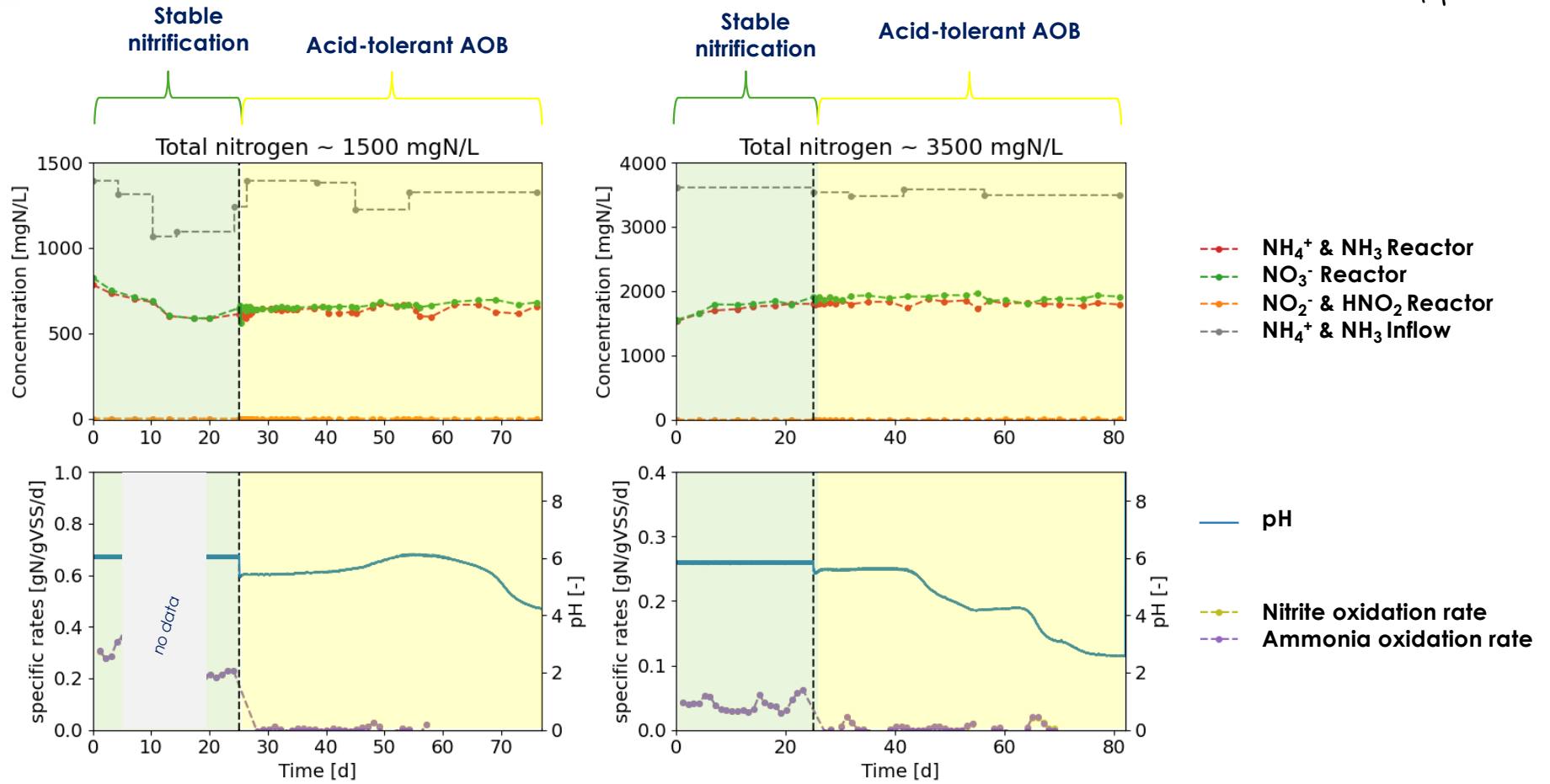
pH=6.0/6.05 and increase to pH=7.0/7.05



pH=6.0/6.05 and increase to pH=8.50/5.55

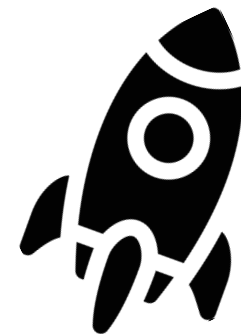
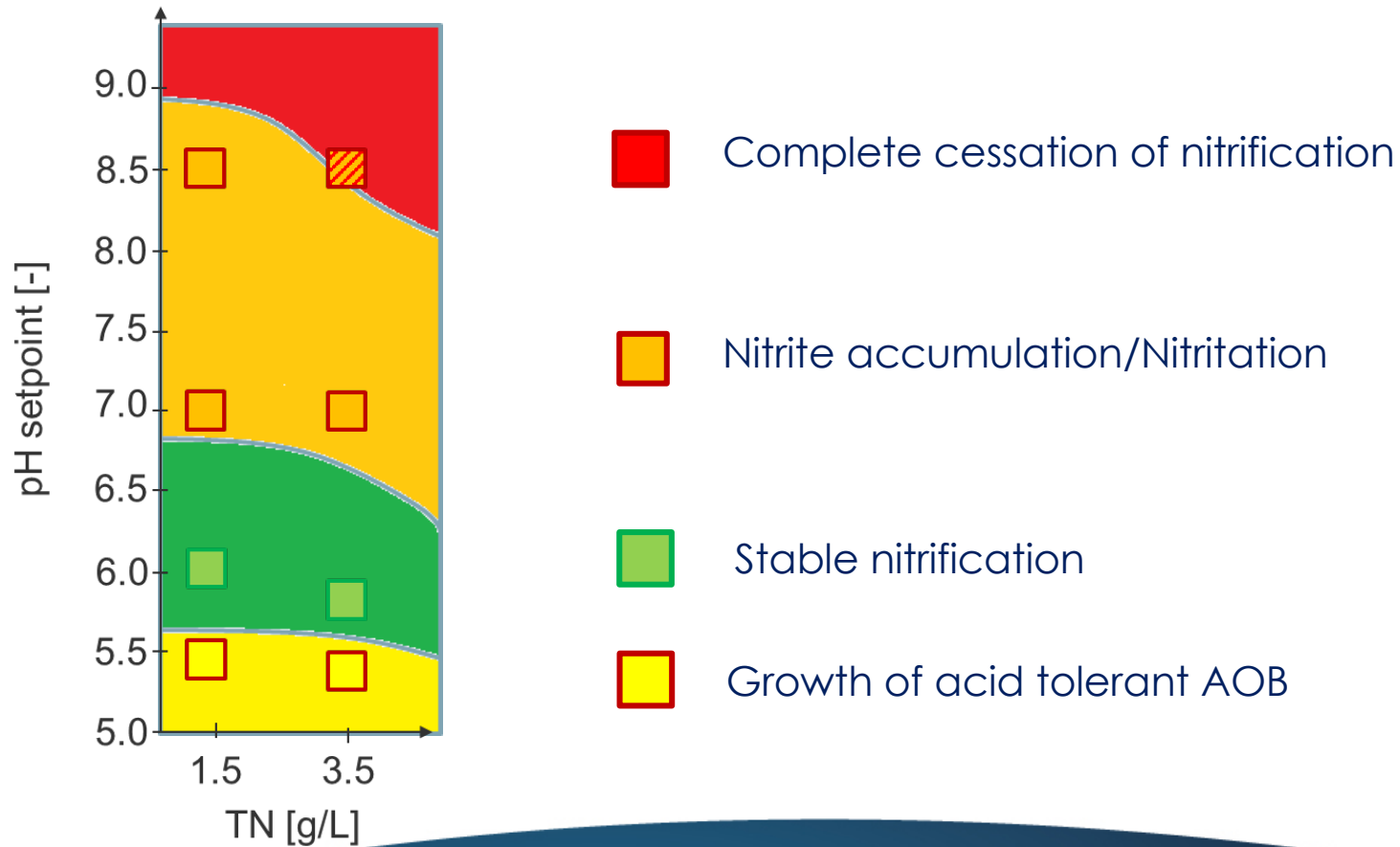


pH=6.0/6.05 resp. pH=5.8/5.85 and stop inflow





Conclusion and outlook



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THANK YOU.

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